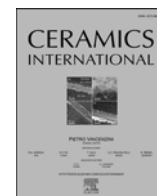




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Significantly enhanced sinterability and temperature stability of $\text{Li}_3\text{Mg}_4\text{NbO}_8$ -based microwave dielectric ceramics with LiF and $\text{Ba}_3(\text{VO}_4)_2$ addition

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ABSTRACT

$\text{Li}_3\text{Mg}_4\text{NbO}_8$ -basic composite ceramics were elaborated via the solid-state reaction process, in which LiF and $\text{Ba}_3(\text{VO}_4)_2$ were utilized as a sintering aid and reinforcement phase, respectively. The sinterability, phase assemblage, microstructures, and microwave dielectric performances of $\text{Li}_3\text{Mg}_4\text{NbO}_8$ -LiF- $\text{Ba}_3(\text{VO}_4)_2$ composite ceramics were thoroughly researched. The co-addition of LiF- $\text{Ba}_3(\text{VO}_4)_2$ can simultaneously lower the sintering temperature and improve the thermal stability of $\text{Li}_3\text{Mg}_4\text{NbO}_8$ -basic ceramics. Solid state activated sintering is responsible for the low-temperature densification of the present ceramics. The coexistence of rock-salt structural $\text{Li}_3\text{Mg}_4\text{NbO}_8/\text{Li}_4\text{Mg}_4\text{NbO}_8\text{F}$ and hexagonal structural $\text{Ba}_3(\text{VO}_4)_2$ phases was demonstrated by the combinational XRD and SEM-EDS analysis results. The $0.65(\text{Li}_3\text{Mg}_4\text{NbO}_8\text{-LiF})\text{-}0.35\text{Ba}_3(\text{VO}_4)_2$ ceramics fired at $825^\circ\text{C}/5\text{ h}$ exhibited promising microwave dielectric performances: $\tau_f = 0.5\text{ ppm}/^\circ\text{C}$ along with $\epsilon_r = 13.8$ and $Q \times f = 68500\text{ GHz}$. The good compatibility of the developed ceramics with Ag demonstrates its potential for use in LTCC technology.

1. Introduction

As the prevalence of 5th generation (5G) wireless communication systems and Internet of Things, microwave dielectric ceramics have drawn special attentions because they are extensively applied as microwave passive devices including substrates of filters and antennas in wireless communication devices [1,2]. Besides, low-temperature co-fired ceramic (LTCC) technology is beneficial for realization the miniaturization and integration of future microwave devices [3–5]. To meet the requirements of 5G and LTCC technologies, dielectric ceramics should have two essential characteristics: (1) good dielectric performances including low permittivity (ϵ_r), low dielectric loss or high quality factor ($Q \times f$), and small or zero temperature coefficient of resonant frequency (τ_f); (2) low firing temperature ($\leq 960^\circ\text{C}$) and excellent compatibility with Ag electrodes [6,7]. At present, numerous researches have been conducted to exploit new microwave dielectric ceramic systems, ameliorate the dielectric properties of specific ceramics via ion substitution or the formation of new composite materials, and reduce the firing temperature for specific ceramics by doping sintering aids [8–11].

Recently, rock-salt structural $\text{Li}_2\text{O-MgO-Nb}_2\text{O}_5$ system ceramics have attracted considerable attentions in commercial and scientific research owing to their fascinating dielectric and luminescence characteristics [12–16]. Our previous work showed that a new compound within the $\text{Li}_2\text{O-MgO-Nb}_2\text{O}_5$ system, namely, $\text{Li}_3\text{Mg}_4\text{NbO}_8$, was fabricated, it exhibited excellent dielectric performances at 9.6 GHz ($\epsilon_r = 13.8$, $Q \times f = 103\,400\text{ GHz}$) [17]. However, the relative large negative τ_f value ($-36\text{ ppm}/^\circ\text{C}$) and high sintering temperature (1150°C) of $\text{Li}_3\text{Mg}_4\text{NbO}_8$ ceramics inhibited its use in LTCC applications. Previous researches demonstrate that LiF is a valid sintering aid for lowering the densification temperature of dielectric ceramics, particularly in Li-containing rock-salt structural ceramics [18–20]. For the abnormal low ϵ_r but positive τ_f values, the $\text{Ba}_3(\text{VO}_4)_2$ is often utilized as τ_f compensator in diphase ceramics such as $\text{Ba}_3(\text{VO}_4)_2\text{-Ca}_5\text{Mg}_4(\text{VO}_4)_6$, $\text{Li}_3\text{Mg}_2\text{SbO}_6\text{-Ba}_3(\text{VO}_4)_2$, etc [21,22].

In this paper, the LiF as sintering aid and $\text{Ba}_3(\text{VO}_4)_2$ as τ_f compensator were introduced to $\text{Li}_3\text{Mg}_4\text{NbO}_8$ ceramics, with the aim to concurrently obtain low sintering temperature and excellent dielectric performances of $\text{Li}_3\text{Mg}_4\text{NbO}_8$ -basic ceramics. Therefore, we designed and prepared the $0.65(\text{Li}_3\text{Mg}_4\text{NbO}_8\text{-LiF})\text{-}0.35\text{Ba}_3(\text{VO}_4)_2$ ceramics based

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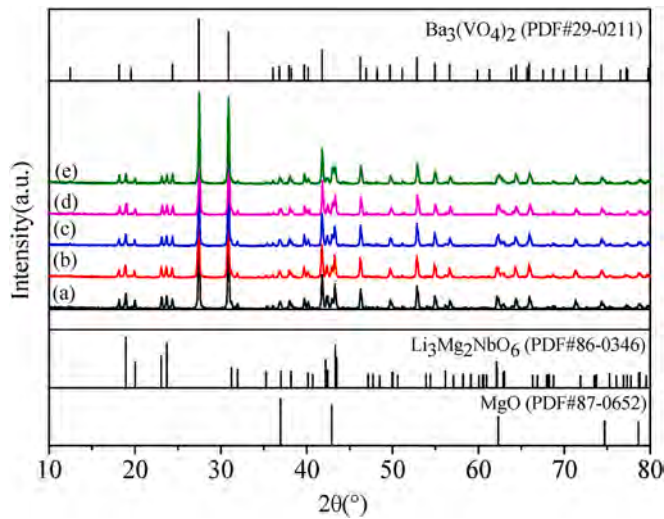


Fig. 1. XRD patterns of LMN-LiF-BV ceramics sintered at 750 °C–850 °C: (a) 750 °C, (b) 775 °C, (c) 800 °C, (d) 825 °C, (e) 850 °C.

on the mixing rule of τ_f and via a routine solid state reaction process [23]. Furthermore, we thoroughly investigated the impacts of $\text{Ba}_3(\text{VO}_4)_2$ and LiF addition on the phase purity, sinterability, microstructures, microwave dielectric characterization, and compatibility with silver electrode of 0.65($\text{Li}_3\text{Mg}_4\text{NbO}_8$ -LiF)-0.35 $\text{Ba}_3(\text{VO}_4)_2$ sintered bodies.

2. Experimental procedure

The 0.65($\text{Li}_3\text{Mg}_4\text{NbO}_8$ -LiF)-0.35 $\text{Ba}_3(\text{VO}_4)_2$ (abbreviated as LMN-LiF-BV) samples were elaborated via a classical solid-state process. The starting powders of MgO, Li_2CO_3 , Nb_2O_5 , BaCO_3 , V_2O_5 (all purity > 99.0 %, Guo-Yao Co. Ltd., Shanghai, China) were individually weighed according to stoichiometric $\text{Li}_3\text{Mg}_4\text{NbO}_8$ and $\text{Ba}_3(\text{VO}_4)_2$, then were separately mixed together for 8 h via planetary milling using zirconia and alcohol as media and solvent. The resultant milled powders of $\text{Li}_3\text{Mg}_4\text{NbO}_8$ and $\text{Ba}_3(\text{VO}_4)_2$ were presintered at 1000 °C/4 h and 650 °C/20 h, respectively. Then the LMN-LiF-BV mixture powders were remilled. The obtained powders were granulated using a 6 wt% PVA solution, and pelletized into green discs ($\Phi 10 \times 5$ mm). These discs were sintered at 750 °C–850 °C for 5 h in air.

The phase constitutions of the LMN-LiF-BV ceramics were examined

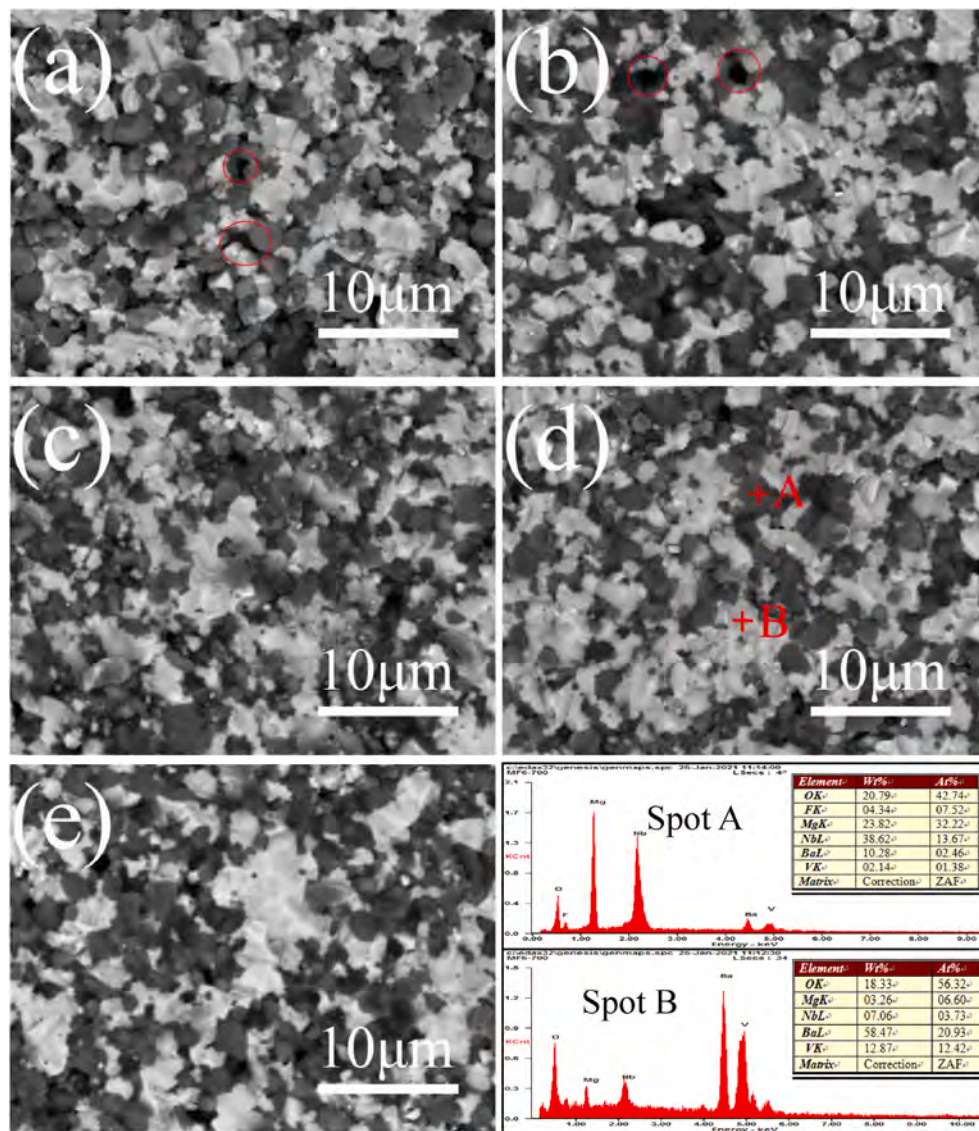


Fig. 2. The typical cross-section SEM images of LMN-LiF-BV ceramics sintered at different temperature: (a) 750 °C, (b) 775 °C, (c) 800 °C, (d) 825 °C, (e) 850 °C, and (f) EDS analysis of Fig. 2(d) points A and B.

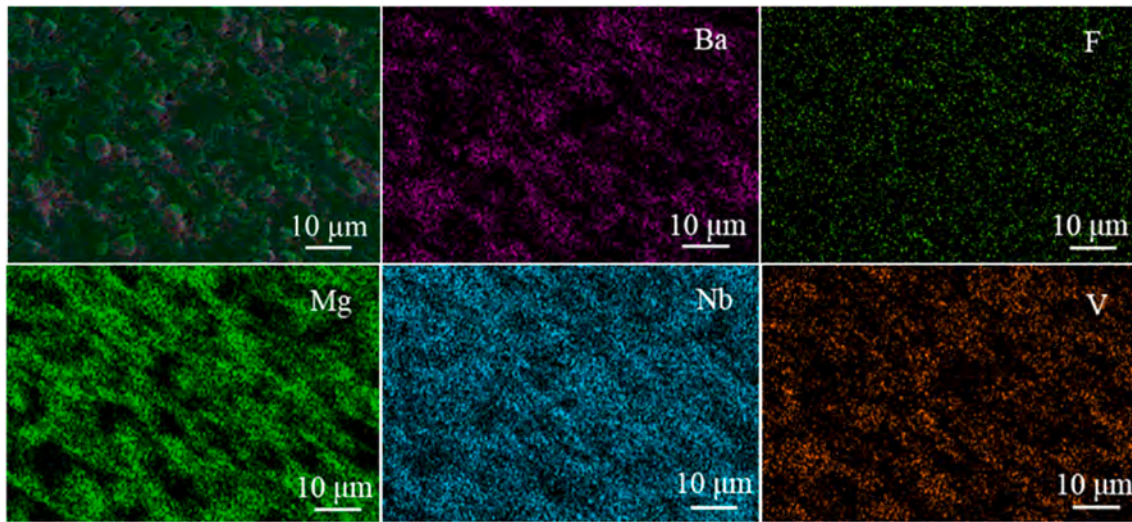


Fig. 3. Element mapping of Ba,V, Mg, Nb, F of LMN-LiF-BV ceramics sintered at 825 °C.

via X-ray powder diffraction (XRD, SmartLab, Japan). The fracture morphologies, and the main chemical composition of specimens were measured using a scanning electron microscope, (SEM, XL-40FEG, Philips) coupled with an energy-dispersive spectrometer (EDS). The apparent densities of sintered bodies were tested using Archimedes' principle. The ϵ_r and $Q \times f$ values of the LMN-LiF-BV ceramics were evaluated using a precision vector network analyzer (N5230A, Agilent, USA) under 9–10 GHz via the resonant cavity method. At 20 °C and 80 °C, the resonance frequency was tested to estimate the τ_f value using the following formula (1):

$$\tau_f = \frac{(f_{80} - f_{20}) \times 10^6}{f_{20} \times (80 - 20)} \quad (1)$$

3. Results and discussion

Fig. 1 presents the XRD profiles of 0.65(Li₃Mg₄NbO₈-LiF)-0.35Ba₃(VO₄)₂ (LMN-LiF-BV) ceramics sintered at 750 °C–850 °C. No significant change was found in the XRD patterns in all sintered bodies. The main diffraction peaks can be attributed to the Ba₃(VO₄)₂ phase (PDF#25-0902), and other diffraction peaks coincide with the orthorhombic Li₃Mg₂NbO₆-like (PDF#25-0902) or cubic MgO-like (PDF#87-0652) rock salt structure phase. The coexistence of cubic and orthorhombic rock salt phases and the absence of an obvious LiF phase indicated that the Li⁺ and F⁻ ions entered the lattice of ordered orthorhombic Li₃Mg₄NbO₈, thus forming disordered cubic Li₄Mg₄NbO₈F. Similar phenomena were reported in other rock salt structure systems [1,24]. In addition, no additional phases emerged for all sintered specimens, except for the Ba₃(VO₄)₂, Li₃Mg₄NbO₈ and Li₄Mg₄NbO₈F phases, thereby indicating the good compatibility of the Li₃Mg₄NbO₈-Ba₃(VO₄)₂ end-member. The good coexistence of Li₃Mg₄NbO₈-Ba₃(VO₄)₂ composite ceramics can be ascribed to their different crystal structures, which is favor of compensating for the dielectric properties of Li₃Mg₄NbO₈-basic ceramics, especially for τ_f [21].

The morphological microstructures of LMN-LiF-BV composite ceramics sintered at different temperatures are presented in Fig. 2. When the specimens were heated at a temperature under 800 °C or above 850 °C, large intergranular pores (as indicated by red circles) or uneven grain-size distribution were observed. As shown in Fig. 2 (d), a relatively compact microstructure with few pores was achieved, which corresponds to the saturated bulk density and dielectric properties (Fig. 3). Meanwhile, all specimens exhibited similar grain sizes and two different-colored grains (bright and dark), which possibly denote distinct phase assemblages [25]. Thus, EDS spectra were performed on

the 825 °C-sintered LMN-LiF-BV ceramics as a representative to assess the phase assemblage for the different-colored grains. As shown in the EDS spectra of Fig. 2(f), the dark grains (spot A) are rich in Mg, F and Nb but relatively poor in Ba, and V, whereas the bright grains (spot B) are rich in Ba and V but relatively poor in Mg, F and Nb. In combination with the XRD results and considering the fact that the Li element is too light to be detected, the dark and bright grains corresponded approximately to Li₃Mg₄NbO₈/Li₄Mg₄NbO₈F and Ba₃(VO₄)₂, respectively. Furthermore, SEM mapping was used to exhibit the element distribution. As revealed by the element mapping in Fig. 3, Mg, Nb and F elements are mainly distributed in the dark grains, whereas Ba and V elements are mainly distributed in the bright grains, respectively. The element mapping results are consistent with the EDS spectra results.

Fig. 4 illustrates the variation of bulk density (ρ) and dielectric performances (ϵ_r , $Q \times f$, τ_f) of LMN-LiF-BV ceramics with respect to sintering temperature. The ρ value rose gradually, and reached the saturated value at approximately 825 °C as increasing sintering temperature. The variations of ρ are consistent with the microstructure of SEM observation. The variation tendency of the ϵ_r values of the LMN-LiF-BV ceramics was the same as that of the ρ because a high ρ corresponds to a high ϵ_r [26]. As shown in Fig. 4(b), the $Q \times f$ value for LMN-LiF-BV ceramics rose initially, reached the saturated value at around 825 °C, and then declined with increasing temperature, although its ρ slightly increased. In general, the $Q \times f$ value of microwave ceramics

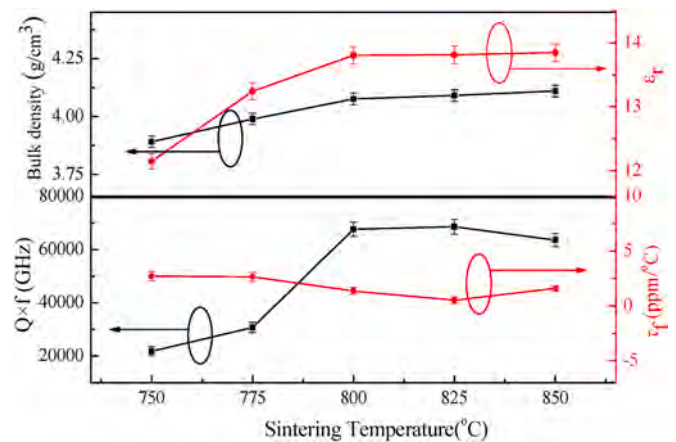


Fig. 4. Dependence of ρ , ϵ_r , $Q \times f$ and τ_f of LMN-LiF-BV ceramics on sintering temperature.

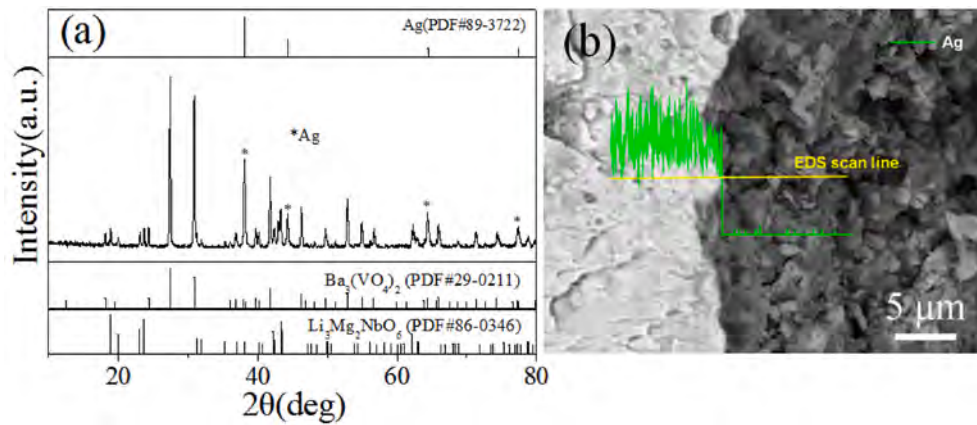


Fig. 5. The XRD pattern, interface morphology and EDS line scan of LMN-LiF-BV-20 wt% Ag sample sintered at 825 °C.

is primarily affected by extrinsic factors, such as impurity phases, bulk density, and defects [27–29]. In our case, the increment in the $Q \times f$ value can be attributed to the improvement of the density, whereas the decline of the $Q \times f$ value is possible because of the defect caused by exaggerated grain growth [30]. As illustrated in Fig. 4(b), the impact of sintering temperature on the τ_f values of LMN-LiF-BV seemed to be small, and the τ_f value remained stable at approximately 1.0 ppm/°C because no remarkable changes were observed in constitutions (Fig. 1). In summary, the optimum dielectric properties were achieved for LMN-LiF-BV ceramics sintered at 825 °C. The densification temperature (825 °C) of the present ceramics is lower than the melting point (850 °C) of LiF [31]. Thus, we infer that the low temperature sintering mechanism of present ceramics is solid-state activated sintering [32]. Specifically, the incorporation of LiF into the $\text{Li}_3\text{Mg}_4\text{NbO}_8$ lattice decreased the chemical potential, thus reducing the firing temperature for the present samples [31–33].

To assess the suitability of LMN-LiF-BV ceramics in LTCC application, its chemical compatibility with Ag electrode was examined. Fig. 5 displays the XRD pattern along with the interface morphology and EDS line scan of LMN-LiF-BV with 20 wt% Ag addition sintered at 825 °C. No other additional phase besides LMN-BV composite ceramics and Ag phases could be obviously detected from the XRD pattern in Fig. 5(a). This indicated that no chemical reaction occurred between them during the sintering process. Furthermore, no delamination and element diffusion phenomena were observed at the interface between the LMN-LiF-BV ceramics and the Ag electrode, as shown in Fig. 5(b). These characters demonstrate that the LMN-LiF-BV ceramics could be well matched with the Ag electrode. Moreover, Table 1 summarizes the microwave dielectric properties and optimal firing temperature of some ceramics in the $\text{Li}_2\text{O-MgO-Nb}_2\text{O}_5$ system. As shown in Table 1, our $\text{Li}_3\text{Mg}_4\text{NbO}_8$ -LiF-BV ceramics exhibit a lower sintering temperature or higher $Q \times f$ value compared with other related ceramics. Thus, the

foreseeable potential of LMN-LiF-BV ceramics for LTCC applications should be expected.

4. Conclusions

New temperature-stable LMN-LiF-BV diphasic ceramics were prepared via the solid-state reaction process. The impacts of LiF- $\text{Ba}_3(\text{VO}_4)_2$ introduction on the phase assemblage, firing behavior, microstructures, and microwave dielectric characteristics of $\text{Li}_3\text{Mg}_4\text{NbO}_8$ -basic ceramics were systematically researched. The remarkably low sintering temperature and favorable thermal stability of $\text{Li}_3\text{Mg}_4\text{NbO}_8$ -basic ceramics can be attributed to the introduction of LiF and $\text{Ba}_3(\text{VO}_4)_2$, respectively. XRD and SEM-EDS analyses confirmed that the rock salt structural ($\text{Li}_3\text{Mg}_4\text{NbO}_8$, $\text{Li}_4\text{Mg}_4\text{NbO}_8\text{F}$) and hexagonal structural $\text{Ba}_3(\text{VO}_4)_2$ phases can stably coexist. Typically, the optimal microwave dielectric characteristics were achieved for LMN-LiF-BV ceramics sintered at 825 °C, where $\epsilon_r = 13.8$, $Q \times f = 68500 \text{ GHz}$ (at 9.4 GHz), and $\tau_f = 0.5 \text{ ppm/}^\circ\text{C}$. Besides, the good compatibility of the present ceramics with Ag further makes it a potential alternative material for LTCC technology.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Table 1

Comparison of the proposed dielectrics with some compounds in $\text{Li}_2\text{O-MgO-Nb}_2\text{O}_5$ system.

Compounds	ϵ_r	$Q \times f$ (GHz)	τ_f (ppm/°C)	ST (°C)	Ref.
$0.65(\text{Li}_3\text{Mg}_4\text{NbO}_8\text{-LiF})$ $-0.35\text{Ba}_3(\text{VO}_4)_2$	13.8	68500	0.5	825	^a
$\text{Li}_3\text{MgNbO}_5\text{-CaTiO}_3$	18.4	86625	1.2	1260	[14]
$0.7\text{Li}_3(\text{Mg}_{0.92}\text{Zn}_{0.08})_2\text{NbO}_6$ $-0.3\text{Ba}_3(\text{VO}_4)_2$	16.3	50084	1.5	950	[34]
$\text{Li}_3\text{Mg}_2\text{NbO}_6\text{-}0.1\text{TiO}_2\text{-LBBS}$	16.0	42648	-1.0	900	[35]
$\text{Li}_3\text{Mg}_2(\text{Nb}_{0.94}\text{W}_{0.06})\text{O}_{6.03}$	15.8	124187	-18.3	1175	[36]
$\text{Li}_3\text{Mg}_2(\text{Nb}_{0.94}\text{Nb}_{0.06})\text{O}_6$	15.8	95487	-24.3	1150	[37]

^a This work.

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